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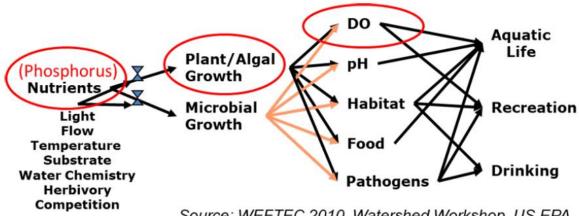




## 1. BACKGROUND

## 1.1 US EPA and Illinois Nutrient Strategy

The efforts to reduce nutrient-related pollution are ongoing in many states. The US EPA pushes for states to develop numeric criteria (NNC). NNC must protect designated use, but the relationship is not linear.



Source: WEFTEC 2010, Watershed Workshop. US EPA.

Illinois nutrient strategy set the goal to reduce total phosphorus (TP and nitrogen (TN) loads by 45% by 2045. Illinois Nutrient Science Advisory Committee developed the instream NNC. However, the Standards were not adopted by Illinois Pollution Control Board. The Environmental groups wanted 0.1 mg/L TP in POTW permits.

In 2018, an Agreement were reached between Illinois Association of Wastewater Agencies, Illinois EPA, and environmental groups, and set the goal for major WWTPs to reach 0.5mg/L TP by 2030. Special conditions were developed in NPDES permits to address of phosphorus related impairments, such as dissolved oxygen (DO) and nuisance algae. The Agreement also allows some flexibility to develop water-shed-specific targets. The Nutrient Assessment and Reduction Plan (NARP) was born.

Based on instream sampling by IEPA, phosphorus related impairment is listed on 303(d) list for DO and offensive condition (algae and/or aquatic plant growth).

## 1.2 IDOC-Dixon Correctional Center WWTP

Dixon Correctional Center WWTP was designed to treat the wastewater generated solely from the entire operations of the correctional center. The major design Influent parameters are as follows (See Appendix A. Basis of Design for details):





- Average Daily Water Flow (ADWF)
- Maximum Daily Waste Flow (MDWF)
- CBOD (Influent)

1.0 mgd (694 gpm) 3.0 mgd (2,082 gpm) 400 mg/l (3.340 lbs/d) 350 mg/l (2,920 lbs/d)

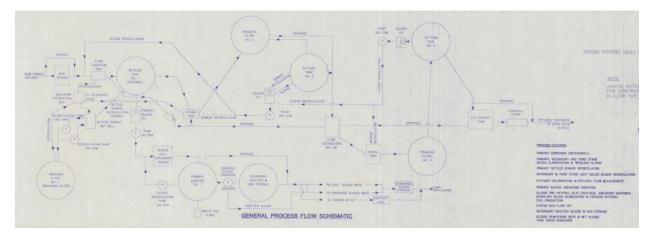
Suspended Solid (Influent)

The effluence is discharged to Rock River through a 24" sewer pipe system.

The treatment plant adopts multi-staged Trickling Filter Biological treatment processes with headwater, primary and secondary settling tanks, chlorine contact tank, etc. for wastewater treatment.

Sludge treatment unit uses anaerobic digestion biological process, with primary digester, secondary digester, sludge dewatering beds, and dewatered sludge pad. The treatment sludge is used for land application. See Figure 1: GENERAL PROCESS FLOW SCHEMATIC for details (Enlarged Schematic is also attached in Appendix B).

Figure 1



The WWTP is operating under NPDES Permit No. IL0024724 (Bureau ID: W103020000) issued on December 27, 2019, expiration date is December 31, 2024. The permitted total Phosphorus (as P) effluent limitation is "monitor only, 1 day/month".

The NPDES Permit No. IL0024724 includes 17 special conditions. In order to renew the NPDES Permit, the Permittee – IDOC Correctional Center is required to address Special Conditions 16 & 17. See Appendix C.1 Special Condition 16 and Appendix C.2 Special Condition 17 for details.

In terms of Special Condition 16, the limit of 0.5 mg/L Total Phosphorus is 12 month rolling geometric mean (calculated monthly) basis is not technologically feasible through the use of biological phosphorus removal (BPR) processes(se) at IDOC – Dixon Correctional Center facility.





Therefore, the current facility won't be able to meet the effluent limit by January 1, 2030 without upgrade the treatment plant, such as adding chemical/physical processes.

The Permittee develops a written plan, preliminary engineering report or facility plan no later than January 1, 2025, to rebuild or replace the secondary treatment process(se) of the treatment facility, the Limit shall be met by December 31, 2035.

In terms of Special Condition 17, the Permittee submits a NARP to the Agency by December 31, 2023. The plan is to meet the 0.5 mg/L limit by December 31, 3035.

## 1.3 Nutrient Assessment Reduction Plan (NARP)

IDOC – Dixon Correctional Center facility developed its own NARP to meet the effluent limit of 0.5 mg/L total Phosphorus by December 31, 2035. The overall strategies are to:

- Join/establish a watershed group
- Develop NARP objective
- Determine P-reduction or other measures to address impairments
- Establish schedule

## 1.4 Join/Establish a Watershed Group

Dixon Correctional Center WWTP was categorized as NARP – Risk. Figure 3. GIS Map of Illinois NARP for all concerned facilities and Watershed Group, see Figure 2: The treatment plant is in Lower Rock River Watershed (HUC 07090005).

See Figure 3: The Watershed Group, includes the following entities:

- 1. Rock River Water Reclamation District (W2010300010)
- 2. IDOC Dixon Correctional Center (W1030200009)
- 3. Dixon WWTP (W1030200001)
- 4. City of Sterling WWTP (W1950500003)
- 5. Rock Falls New STP (W1958090002)
- 6. Moline South ST (W1610450003)
- 7. Village of Poplar Grove South STP (W0070150006)
- 8. Belvidere WWTP (W0070050001)
- 9. City of South Beloit STW (W2010450005)
- 10. City of Marengo STW (W1110650003)

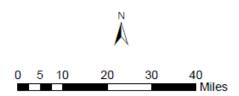
IDOC – Dixon Correctional Center reached out to Rock River Watershed Group, but was told that the Group no longer accepted any new group members. See Table 1 Rock River Watershed Group Board Contact List,

IDOC - Dixon Correctional Center WWTP will develop its own Nutrient Assessment Reduction Plan (NARP) that will meet the requirements.





## Lower Rock Watershed (HUC 07090005)



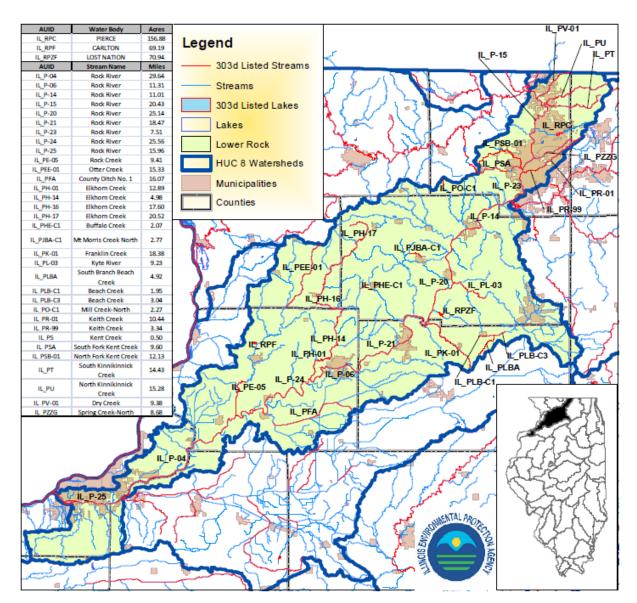


Figure 2 Lower Rock River Watershed (HUC 07090005)





Table 1
Rock River Watershed Group Board Contact List

	Rochelle	
Adam Lanning	alanning@rmu.net	815-564-7174
Adriana Milan	amilan@rochelleil.us	
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-		815-353-5839
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	Four Rivers Sanitation Authority	
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	Black & Veatch	
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Brian Valleskey	Bvalleskey@geosyntec.com	
Rishab Mahajan	Rmahajan@geosyntec.com	

Updated 4/18/22 by MO





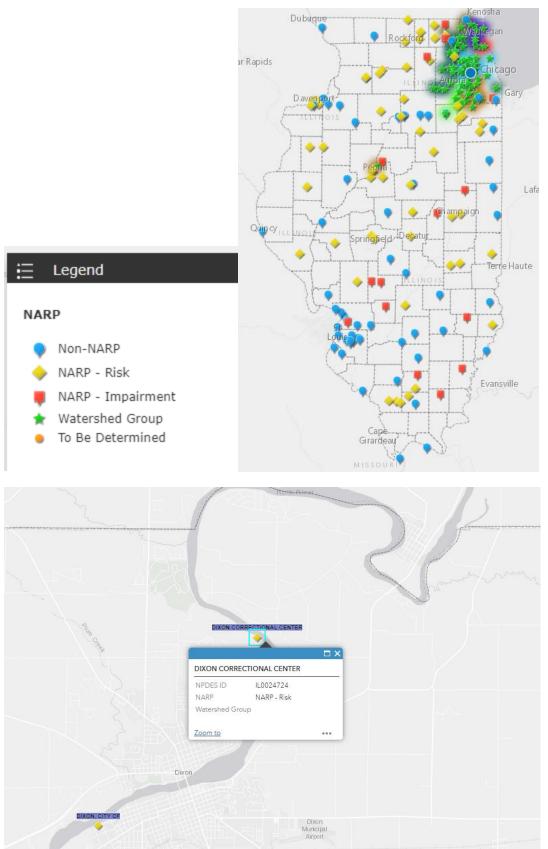


Figure 3. GIS Map of Illinois NARP for all concerned facilities and Watershed Group





## 2. NARP OBJECTIVES

The NARP developed by IDOC – Dixon Correctional Center facility is to meet the effluent limit of 0.5 mg/L total Phosphorus by December 31, 2035. The Permittee will develop written plan, preliminary engineering report or facility plan no later than January 1, 2025, to rebuild or replace the secondary treatment process(se) of the treatment facility, the Limit shall be met by December 31, 2035.

### 2.1 Determine P-Reductions or Other Measures

### 2.1.1 Dixon Facility monthly monitoring

IDOC Dixon Facility compiled 12-month rolling total Phosphorus effluent concentrations between April 2022 and March 2023. The 12-month rolling geometric mean (calculated monthly) is 2.13 mg/L. See <u>Appendix D 12-Month Rolling TP Monitored in Effluent Flow</u>. The Permittee also investigated the influent sources of Phosphorus. A sample was collected on June 8<sup>th</sup>, 2023 from influent flow, and analyzed on June 19<sup>th</sup>, 2023. The total Phosphorus was 6.57 mg/L. See <u>Appendix E Analytical Report of TP in Influent Flow</u>.

### 2.1.2 Phosphorus Source Investigation

In addition to normal sources of phosphorus from sanitary sewerage generated by the institutional facility, IDOC Dixon facility uses CARUS<sup>™</sup> 8600 water treatment chemical to inhibits corrosion of lead and copper plumbing for the public water supplying systems. The chemical residues eventually enter the wastewater treatment plant.

The boiler house also uses a phosphate/polymer blend in the boiler for corrosion control, however very little of this product is discharged to the treatment facility.

The Engineer of the Permittee investigated the non-phosphorus based chemicals for inhabitance of corrosion purposes, such as FlexPro<sup>®</sup> CL and other chemicals, but failed to find any suitable substitutions.

### 2.1.3 Total Phosphorus Removal Processes – Scientific Supports

The treatment plant adopts multi-staged Trickling Filter Biological treatment processes with headwater, primary and secondary settling tanks, chlorine contact tank, etc. for wastewater treatment. Because the trickling filter biological treatment processes do not have flexibility of aeration re-arrangement to modify, the enhanced biological treatment processes may not apply to IDOC Dixon facility. Instead, most likely chemical/physical treatment processes will be investigated and implemented to meet the goal of effluent limit of 0.5 mg/J TP.



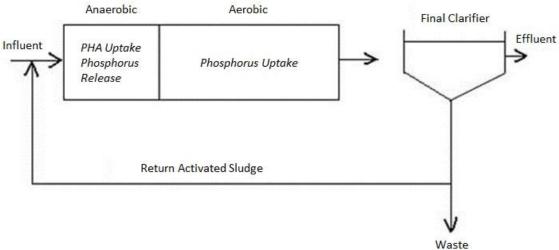


There are two major total phosphorus removal processes. They are Biological Removal Process and Chemical Process. The Permittee will look into both removal processes and select the best available solutions.

### Biological Removal – Enhanced Biological Phosphorus Removal (EBPR)

EBPR relies on the selection and proliferation of a microbial population capable of uptaking orthophosphate in greater amounts than their normal biological growth requirements. EBPR is a process that uses alternating anaerobic and aerobic zones to provide an environment that encourages the growth of phosphorus-accumulating organisms (PAO). PAOs store excess polyphosphate in their cell mass and phosphorus is removed with the waste sludge (see figure). Graphic source: Jeremy Cramer, Stevens Point

## A typical EBPR reactor configuration

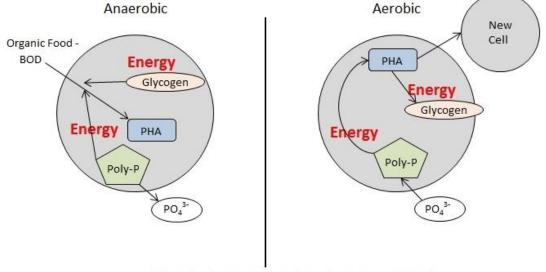








The unique feature of EBPR is the anaerobic selector used in the treatment process (See figure)



## Mechanism of BPR

Phosphorus Accumulating Organisms (PAOs)

PAOs use polyphosphate (poly-P) and glycogen stored in their cells as energy sources to enable them to uptake volatile fatty acids (VFA). VFAs are converted to polyhydroxyalkanoates (PHA) and stored in the cells of PAOs. As they take up VFA, the PAOs release orthophosphate into the mixed liquor. PAOs do not grow in the anaerobic zone but their ability to uptake food in the form of VFAs gives them a competitive advantage over other bacteria.

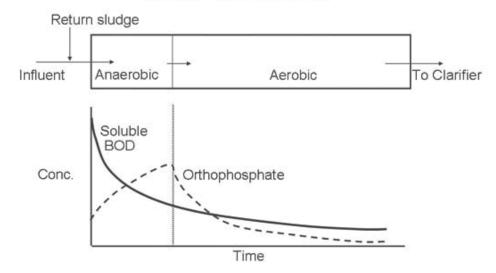
In the aerobic zone, PAOs use PHA as a source of carbon and energy for metabolism and cell growth. PAOs will also restore their supplies of glycogen and polyphosphate in the aerobic zone. To replenish their stored polyphosphate, PAOs will take up excess phosphate from the mixed liquor, the mechanism of EBPR.

In the anaerobic zone, PAOs will rapidly take up BOD (as VFAs) and release orthophosphate into the mixed liquor. As the wastewater passes through the anaerobic zone, VFA will rapidly decrease and orthophosphate will increase. In the aerobic zone, BOD will continue to decrease. As PAOs restore their polyphosphate supplies in the aerobic zone, the concentration of orthophosphate in the mixed liquor will rapidly decrease (see figure).

Graphic source: Jenchie Wang, Symbiont

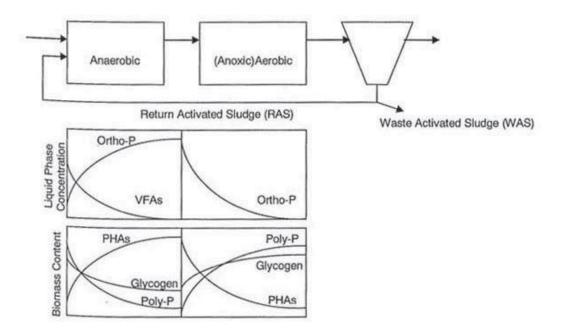






## Profile of BOD and P in mixed liquor as it passes through an EBPR plant

Under anaerobic conditions, PAOs take up VFA from the mixed liquor and store it as PHA within their cells. To do this, PAOs use the glycogen and polyphosphate as energy sources; therefore depleting their stores of these compounds. Under aerobic conditions, PAOs use up their stored PHA for metabolism and growth and to restock their supplies of glycogen and polyphosphate. To build up their supply of polyphosphate, PAOs will take up excess orthophosphate from the mixed liquor in the aerobic zone.

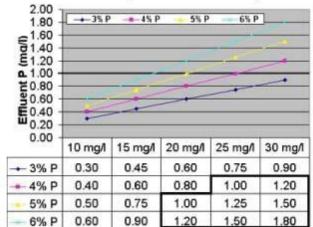






Under anaerobic conditions, PAOs are at a competitive advantage to uptake a readily available food source (VFAs) and therefore are selected for in this environment. Most other bacteria cannot uptake the VFAs under anaerobic conditions.

Sludge phosphorus content is defined as the percentage of phosphorus in cell mass and is expressed as phosphorus/volatile suspended solids (VSS)%. In a conventional activated sludge treatment process, the sludge phosphorus content is approximately 1.5% to 2.5%. In an EBPR system, the sludge phosphorus content is 3.0% to 6.0% or higher. The more efficient the EBPR system is, the higher the sludge phosphorus content will be. Phosphorus is removed from the treatment system by wasting sludge. Because the activated sludge phosphorus content is high in an EBPR plant, effluent TSS should be kept low. A small amount of TSS with high phosphorus content could contribute to a high total phosphorus concentration in the effluent.



### Effluent TSS and MLSS %P Relationship on Effluent Phosphorus

Secondary phosphorus release can be defined as phosphorus released from a cell which is not associated with intercellular energy storage. In other words, the secondary phosphorus released is not able to be taken up by the PAOs, which results in a higher phosphorus concentration in the liquid phase, and a reduced phosphorus removal efficiency.

The primary causes of secondary phosphorus release are:

- 1. The retention time of the anaerobic selector of an EBPR process is too long
- 2. The retention time of settled sludge in the clarifier is too long
- 3. The retention time of the aerobic stage of an EBPR process is too long, causing cell lysis and phosphorus release
- 4. The long storage of waste sludge causing phosphorus to be released back into solution and then returned back into the plant through sidestreams





The idea of EBPR is to create conditions in secondary treatment that result in the microorganisms absorbing excess phosphorus. This phosphorus is removed from the flow when the activated sludge is wasted. If a sidestream is high in phosphorus, the sidestream can be treated. Metal salts can be added to these sidestreams to precipitate the phosphorus to avoid overloading the plant. If only the sidestream is treated chemically, biological phosphorus effluent limits still apply.

Problem	Cause	Corrective Action
Biochemical oxygen	High phosphorus in	Monitor the BOD, soluble BOD, TP, and
demand (BOD)/total	sidestream recycles from	orthophosphate in influent to anaerobic zone;
phosphorus (TP) ratio	sludge handling	control volume of sidestream phosphorus recycles
has changed		or provide sidestream phosphorus removal
	Increased phosphorus in raw	Monitor the high TP in raw influent; control
	influent from industrial or	industrial or commercial phosphorus discharge to
	commercial discharges	sanitary sewer system
Phosphorus release is	Insufficient volatile fatty	Monitor VFA/soluble BOD entering anaerobic zone
poor or not occurring in	acids (VFA)	and orthophosphate at end of anaerobic zone;
anaerobic zone		supplement VFA by chemical addition; increase hydraulic retention time (HRT) of anaerobic zone if possible to ferment BOD
Anaerobic zone not truly	Excess dissolved oxygen (DO)	Monitor oxidation reduction potential (ORP) above
anaerobic	from recycle flows	-100 mV in anaerobic zone, DO in recycle; reduce D
		in aeration basin; reduce RAS or internal mixed
		liquor recycle to anaerobic zone
	Excess NO <sub>3</sub> from recycle	Monitor ORP above -100 mV in anaerobic zone and
	flows	nitrate in recycles; reduced RAS or internal mixed
	10005	liquor recycle to anaerobic zone; increase anoxic
		zone HRT if possible to promote denitrification
	Air entrainment from excess	Monitor ORP above -100 mV in anaerobic zone but
	turbulence in anaerobic zone	no nitrates or DO in recycles; reduce turbulence if
	or upstream processes, i.e.	possible
	aerated grit tanks	• 322.3.4969 122. C
Insufficient VFA in	Changes in influent waste	Monitor BOD, soluble BOD, TP and orthophosphate
anaerobic zone	strength	in influent to anaerobic zone; supplement VFA wit
	Ū	chemical addition; add fermented primary sludge
Rapid phosphorus	Secondary release occurring	Monitor TP profile of aeration basin; reduce solids
uptake in aeration basin		retention time (SRT) by wasting more sludge
but effluent TP is higher		
	Secondary release occurring	Check phosphorus in RAS and sludge blanket in fina
	in sludge blanket in final	clarifier; reduce sludge blanket depth
	clarifier	danner) reader staabe stanner acpris
	If phosphorus increases at	Monitor phosphorus profile through anoxic zone;
	end of anoxic zone,	reduce anoxic zone HRT if possible
	secondary release occurring	reduce anoxic zone mixt in possible
	with excessive HRT	
Good phosphorus	If anaerobic HRT is too long,	Monitor TP profile through selector basins; increas
release in anaerobic	secondary release could	RAS to reduce anaerobic HRT
zone but poor	occur after VFAs are used up	NAS LO FERQUE ANALISMU FINI
Standard Andrews	occur arter vr As are used up	
phosphorus removal		





### Chemical Removal Process

The following is the list of the most common chemicals (metal salts) used for phosphorus removal:

- A. Ferric chloride
- B. Ferrous chloride
- C. Ferrous sulfate
- D. Aluminum sulfate (alum)

The characteristics of the chemical (metal salts) used for phosphorus removal

- A. Ferric chloride (FeCl3)
  - 1. Acidic (may lower pH) and alkalinity
  - 2. Very corrosive
  - 3. Fume producing
  - 4. Supplied as 33% to 36% solution (11% to 13% iron)
  - 5. Several grades available
  - 6. Stains concrete and other materials
  - 7. Can affect ultraviolet (UV) disinfection
  - 8. May affect effluent chloride
- B. Ferrous chloride (FeCl2)
  - 1. Acidic (may lower pH) and alkalinity
  - 2. Very corrosive
  - 3. Fume producing
  - 4. Supplied as 18% to 28% solution (8% to 13% iron)
  - 5. Stains concrete and other materials
  - 6. May affect effluent chloride
  - 7. May affect UV disinfection
  - 8. Potential impurities
- C. Ferrous sulfate (FeSO4)
  - 1. Acidic (may lower pH) and alkalinity
  - 2. Very corrosive
  - 3. Fume producing
  - 4. Supplied as 23% to 25% solution (5% to 7% iron)
  - 5. Stains concrete and other materials
  - 6. Should be stored in indoor heated space
  - 7. May affect UV disinfection
  - 8. Potential impurities





- D. Alum (Al2(SO4)3)
  - 1. Acidic (may lower pH) and alkalinity
  - 2. Moderately corrosive
  - 3. 49% aluminum sulfate (8% to 9% aluminum) in liquid form
  - 4. Also available in dry form (powder); must be mixed with water before use
  - 5. Very temperature sensitive in liquid form (must be kept above freezing)
  - 6. Clear, light green, or yellow liquid

When metal ions, iron or aluminum, are added to wastewater two primary precipitates form an insoluble metal phosphate and an insoluble metal hydroxide. For a given metal, the formation of these precipitates is governed by the wastewater alkalinity and soluble orthophosphate concentration in the wastewater, as well as their equilibrium solubility at a given pH.

Polymers can be used as a supplement to enhance phosphorus removal by improving coagulation and settling. Polymers are usually added prior to final clarifiers. The better the settling, the less solids and phosphorus there will be in the final effluent.

As the percent of product increases, the temperature at which it crystallizes will increase.

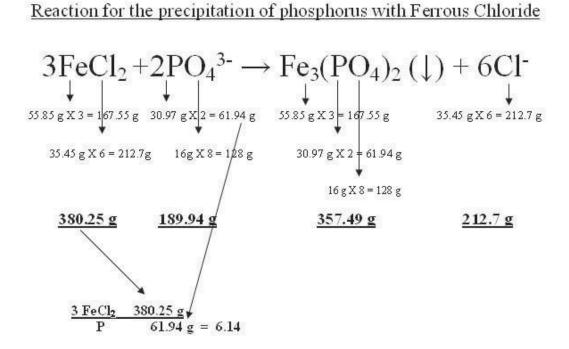
- Ferric chloride (FeCl3)
   42°F at 35% solution
   20°F at 42% solution
- Ferrous chloride (FeCl2) 28°F at 25% Solution 42°F at 35% Solution
- Ferrous sulfate (FeSO4) 42°F at 25% Solution
- Alum (Al2(SO4)3)
   32°F at 8% aluminum sulfate

Crystallization is to be avoided because it is difficult to re-dissolve, plugs pipes and equipment, and forms a solid layer in storage tanks making its removal difficult. Storage room temperatures should be kept warm enough to avoid crystallization.



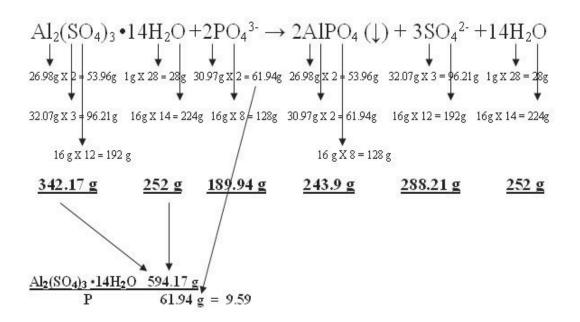


The chemical reaction of the ferrous chloride with phosphorus: Graphic source: Jeremy Cramer, Stevens Point



The chemical reaction of the alum with phosphorus: Graphic source: Jeremy Cramer, Stevens Point

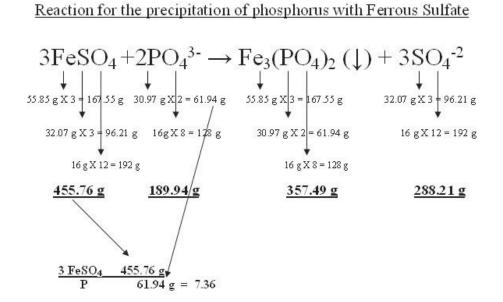
## Reaction for the precipitation of phosphorus with Alum







The chemical reaction of ferrous sulfate with phosphorus: Graphic source: Jeremy Cramer, Stevens Point



Summarize the information of the chemical (metal salts) used of phosphorus removal: Graphic source: Jenchie Wang, Symbiont

Chemical Compound	Ferric Chloride	Aluminum Sulfate (Alum)
A. Formula	FeCl <sub>3</sub>	Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> • 14H <sub>2</sub> O
B. Molecular Weight (g/mole)	162.5	594
C. Typical Reaction With Phosphorus	$FeCl_3 + PO_4^{-3}$ $\rightarrow FePO_4 \downarrow + 3Cl^{-3}$	Al <sub>2</sub> (SO) <sub>3</sub> -14 H <sub>2</sub> O+ 2PO <sub>4</sub> <sup>-3</sup> → 2AIPO <sub>4</sub> ↓ + 3SO <sub>4</sub> <sup>-2</sup> + 14 H <sub>2</sub> O
D: Mole Ratio (Metal : P)	1:1	1:1
E. Weight Ratio (Metal: P)	1.8 : 1	0.87 : 1
F. Mole Ratio (Metal Compound : P)	1:1	1:2
G. Weight Ratio (Metal Compound : P)	5.2:1	9.6:1
H. Commercial Strength in Solution	33 - 36%	48 - 50%
I. Percent of Active Ingredient (Metal)	11 - 13%	8 - 9%
J. Specific Gravity	1.37 @ 35%	1.335 @ 48.5%
K. Amount (Gallons) of Commercial Strength Chemical Solution Required to Remove 1 Pound of Phosphorus Based on 1x Stoichiometric Reaction	1.26 ~ 1.38	1.73 ~ 1.80





Chemical Compound	Ferrous Chloride	Ferrous Sulfate
A. Formula	FeCl <sub>2</sub>	Fe SO <sub>4</sub> • 7H <sub>2</sub> O
B. Molecular Weight (g/mole)	127	278
C. Typical Reaction With Phosphorus	3FeCl <sub>2</sub> + 2PO <sub>4</sub> <sup>-3</sup> $\rightarrow$ Fe <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub> $\downarrow$ + 6Cl <sup>-</sup>	3FeSO <sub>4</sub> 7H <sub>2</sub> O+ 2PO <sub>4</sub> <sup>-3</sup> → Fe <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub> ↓ + 3SO <sub>4</sub> <sup>-2</sup>
D: Mole Ratio (Metal : P)	3:2	3:2
E. Weight Ratio (Metal: P)	2.7 : 1	2.7 : 1
F. Mole Ratio (Metal Compound : P)	3:2	3:2
G. Weight Ratio (Metal Compound : P)	6.1:1	13.5 :1
H. Commercial Strength in Solution	18 - 28%	23 - 25%
I. Percent of Active Ingredient (Metal)	8 - 13%	5 - 7%
J. Specific Gravity	1.40 @ 25%	1.140 @ 25%
K. Amount (Gallons) of Commercial Strength Chemical Solution Required to Remove 1 Pound of Phosphorus Based on 1x Stoichiometric Reaction	1.86 ~ 2.90	5.68 ~ 6.17

Each phosphorus removal chemical (metal salt) has an optimum pH range for precipitating out the phosphorus as a metal phosphate. Iron phosphate [FePO4] and alum [AIPO4] are least soluble at a pH of 6.8 to 7.0, thus precipitate out best at this pH range. Wastewater pH levels outside of this optimum range will require more chemical to achieve the same removal efficiency. While not all wastewaters are at a pH of 6.8 to 7.0, metal phosphates still precipitate out well in the pH range of most wastewaters of 6.0 to 8.5.

Sulfide will react with iron forming a black precipitate. High sulfide wastewater will require higher dosages of iron salts. Sources of sulfides include hauled and certain industrial wastes and collection systems with long detention times.

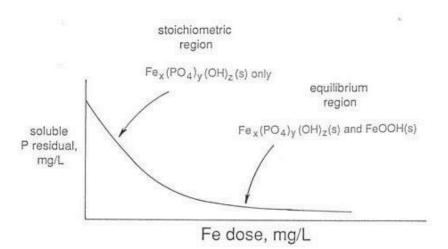
Reaching very low effluent phosphorus concentrations requires increasing amounts of metal salt because of the competition between phosphate and hydroxide precipitation. Using either aluminum or iron salts, a graph of metal dose versus residual dissolved phosphorus can be drawn that illustrates this relationship (see figure 6.3.12.1).

At a given pH and alkalinity, the dissolved orthophosphate concentration in the wastewater will determine whether metal phosphates or metal hydroxides are formed. As the dissolved phosphorus concentration (effluent phosphorus) decreases, more metal hydroxides will be formed.

To achieve low effluent phosphorus limits, increasingly larger doses of metal salts are required to remove additional phosphorus. Eventually, chemical equilibrium will be reached with no further reduction in phosphorus.







Typical Fe dose versus soluble P residual curve.

Actual chemical usage depends on the competing reactions and wastewater characteristics such as pH, alkalinity, and very fine particulate materials (colloids). Wastewater characteristics and competing chemical reactions in the wastewater between the metal salt and phosphorus will result in the need for increased metal salt addition above what was calculated. Biological removal of phosphorus in upstream processes could result in a decreased amount of metal salt addition than calculated. Sampling phosphorus concentrations just upstream of the chemical dose point will help in fine tuning chemical feed rates

Good mixing ensures uniform dispersion of metal salts and efficient chemical use. Overdosing with the metal salt can partially compensate for poor mixing.

Ideally, high intensity mixing at the dose point would be followed by a mixed flocculation zone. However, few treatment plants are designed with rapid mix basins. More typically, metal salts are added at locations where turbulence occurs such as at pump suction lines, mechanical or aerated grit basins, flow-splitting structures, aerated flow-distribution channels, or at hydraulic jumps in Parshall flumes.

A flocculation zone should provide sufficient detention time (15 to 20 minutes) to complete the reaction. Gentle mixing promotes flocculation. The enlarged center feed well on a flocculating clarifier provides such a flocculation zone. Where a flocculation zone is not provided, metal salts should be added far enough upstream of a clarifier to provide adequate reaction time.





Metal salt additives require good mixing to provide effective chemical contact with phosphorus. The dose point may depend on facility design and mixing capability.

- A. Prior to primary clarification
- 1. Advantages
  - a. Metal salt addition upstream of primary clarifiers enhances suspended solids and
  - b. Biochemical oxygen demand (BOD) removal. This reduces loading to the aeration basin
  - c. Odor control (especially iron salts)
  - d. Convenience and mixing capability
  - e. Iron compounds may help balance nutrients in anaerobic digester
  - f. May reduce struvite (phosphate mineral crystal) formation
- 2. Disadvantages
  - a. Greater chemical usage
  - b. May require additional chemical addition downstream
  - c. May result in higher biosolids production
  - d. Could affect nutrient ratio and pH for biological treatment
  - e. Target dosages can be variable due to competing reactions in the primary clarifier and secondary release in the biological treatment system
- B. Addition to secondary treatment (aeration basins)
- 1. Advantages
  - a. Good mixing in aerated basins or lagoons
  - b. Fe+<sup>2</sup> is oxidized to Fe+<sup>3</sup> thus improving reaction with phosphate (PO4<sup>-3</sup>)
  - c. Orthophosphates are readily available
  - d. Convenience
- 2. Disadvantages
  - a. Not recommended for attached growth systems
  - b. Could affect pH, alkalinity, and nutrient ratio
  - c. Not recommended for unmixed ponds and lagoons
  - d. Iron discoloration of structures and equipment
- C. Prior to final clarification
- 1. Advantages
  - a. Efficient and high level of removal because most phosphorus is the orthophosphate form
  - b. Enhances final settling
  - c. Less interfering or competing reactions
  - d. Cost effectiveness
- 2. Disadvantages
  - a. Inadequate mixing





## **3. IMPLEMENTATION SCHEDULE**

The overall schedule is to develop and submit the NARP to IEPA by December 31, 2023, to develop a written plan, preliminary engineering report of the facility plan no later than January 1, 2030, to modify or upgrade the treatment facility, the limit shall be met by December 31, 2035.

- NARP by December 31, 2023
- Written Plan, Preliminary Engineering Report January 1, 2030 (Detailed tasks, budget, and timeframes to be added)
  - 1. Establish the scope of work and budget for the surveying the best available solutions and costs
  - 2. Written plan
  - 3. Survey the costs of the treatment processes
  - 4. Preliminary Engineering Report
  - 5. Establish the total budget commitment to the compliance
- <u>Compliance with 0.5 mg/L Total Phosphorus effluent limit December 31, 2035</u> (Detailed tasks, budget, and timeframes to be added)
  - 1. Engineering Design
  - 2. Construction Plans and Specifications
  - 3. Permit Applications
  - 4. Bid Processes and Contract Award
  - 5. Construction Period
  - 6. Final Inspections
  - 7. Commissioning
  - 8. Operating





## APPENDIX A BASIS OF DESIGN





# APPENDIX B GENERAL PROCESS FLOW SCHEMATIC





# APPENDIX C.1 SPECIAL CONDITION 16





# APPENDIX C.2 SPECIAL CONDITION 17





# APPENDIX D 12-MONTH ROLLING TP MONITORED IN EFFLUENT FLOW





# APPENDIX E ANALYTICAL REPORT OF TP IN INFLUENT FLOW



### DIXON CORRECTIONAL FACILITY WASTEWATER TREATMENT PLANT UPGRADE Dixon, Illinois BASIS OF DESIGN

### I. <u>DESIGN LOADING SUMMARY</u>:

From Discharge Monitoring Reports 10/89 - 5/93 (See attached summary table):

Average Flow	=	.347 mgd
Average Maximum Flow	=	.466 mgd
Maximum Flow in Record	=	.805 mgd (2/93)
Avg. CBOD		379.7 mg/l (1084.3 lbs/d)
Avg. Suspended Solids	=	324.8 mg/l (946.8 lbs/d)

### Proposed Design:

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Average Daily Waste Flow (ADWF)	1.0 mgd	(694 gpm) (1.547 cfs)
Maximum Daily Waste Flow (MDWF)	3.0 mgd	(2082 gpm) (4.64 cfs)
CBOD (Influent)	400 mg/l (3340	lbs/d)
Suspended Solids (Influent)	350 mg/l (2920	lbs/d)

Existing Plant Influent Sewer (24") at .10% (from as-built drawings)

Capacity = 4.6 mgd

Velocity at ADWF =  $\pm 2$  fps

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Dixon WWTP Upgrade Discharge Monitoring Report Summary Tabulation

Chlorine Residual	0.09 0.08 0.08	0.08	0.07	0.11	0.10	0.09	0.07	0.08	70.0	70.0	01.0	0.08	0.10	80'0 1	0.12	0.12	0.00	0.09	0.07	0.08	0.09	0.07	0.09	0.09	0.08	0.09	0.08	0.07	0.10	0.07	0.09	0.07	0.09	0.06	0.06	0.06	0.08	
С.	340 340 75	140	130	<100	×100	<100	<100	010 100	~100 ~100	1/4	~100 ~100	<u>13</u> 2		<pre>&gt;100</pre>	00L>	00 10 10 10 10		0100 V	<100 100</th <th>&lt;100 100</th> <th>&lt;100</th> <th>&lt;100</th> <th>&lt;100 100</th> <th>~100</th> <th>&lt;100 100</th> <th>&lt;100</th> <th>v100</th> <th>&lt;100 100</th> <th>&lt;100 100</th> <th>&lt;100</th> <th>&lt;100</th> <th>×100</th> <th>&lt;100</th> <th>&lt;100</th> <th>&lt;100</th> <th>100</th> <th>&lt;100</th> <th>, , ,</th>	<100 100	<100	<100	<100 100	~100	<100 100	<100	v100	<100 100	<100 100	<100	<100	×100	<100	<100	<100	100	<100	, , ,
Hd	7.4 7.6 7.6	2.6	7.7	7.6	7.7	7.7	7.6	7.6	7.7	7.8	7.9	7.9	χ	ŝ	œ	ω <sup>(</sup>	7.8	7.8	7.8	7.8	7.9	7.9	7.8	7.8	7.7	7.7	7.7	7.7	7.8	7.8	7.8	7.6	7.6	7.7	7.6	7.6	7.6	?
SS Effluent mg/l	10.0 15.1	13.8	7.6	9.1	12.9	14.7	11.4	13.1	11.8	8.4	10.1	<u>6</u> 5	10.8	13.4	10.0	11.8	16.7	14.5	8.7	8.2	6.8	9.5	8.2	7.4	<u>9.0</u>	8.5	7.4	8.2	5.4	4.7	7.8	13.2	8,8	9.8	74	94	.ç	1.2
SS Effluent Ib/d	27.6 39.8	36.0 26.0	20.7	27.6	40.3	50.0	40.5	42.6	39.8	27.3	29.2	25.6	25.5	30.7	26.5	31.8	47.6	45.1	29.2	27.4	22.9	28.3	22.0	17.5	23.5	21.6	18.9	21.4	14.4	11.8	17.1	35.8	24.5	32.4	040	012	0 C C	t. 00
CBOD Effluent mg/l	19.9 19.9	9.7L 8.0L	11.6		16.1	16.0	14.7	12.3	9.8	11.4	10.3	11.8	12.4	14.5	9.6	8.6	8.8	6.9	5.9	5.3	5.0	6.5	8.8	11.3	11.7	12.7	16.1	12.3	7.6	6.2	7.7	7.8	87	13.4	α			13.6
CBOD Effluent Ib/d	54.9 52.4	42.7 52.0	0.00 24 FC	33.7	50.4	54.4	52.2	40.0	33.0	37.1	29.8	32.9	29.3	33.3	25.5	23.2	25.1	21.5	19.8	17.7	16.8	19.4	23.6	26.7	30.5	32.2	41.1	30.1	203	15.4	0.4	0.2 0 0	040	44.1		0. 1 1 1 1 1		4.72
SS Influent Ib/d	1225.1 707.5	640.4 204 4	1.407 040 B	1004.0	0620	1346.0	2106.5	841.4	1708.9	914.3	904.9	678.0	977.4	1060.3	633.9	647.6	764.7	933.6	1260.7	916.3	6 508	744.3	675.4	7647		550.0	2.200	775.0	1116.0	048.0	587.0	0.000	2000	0.759				1379.0
SS Influent mg/l	443.8 302.6	268.5	203.0	0.000	307 B	202.700	2000	258.7	507.2	281.1	312.7	243.4	414.1	462.3	239.0	240.4	268.1	300.1	375 1	274.0	255 1	250.0	201 J	0.1.02	0.012	01210	0800			0.014	0.510	0,102	001 OC		0.000	0.282	340.0	462.0
CBOD Influent Ib/d	1395.7 1074.7	926.2	1014.6	4.799	1210.3	1001 7	004.7 0211 5	11221	1662.1	1256.5	1063.5	1102.0	1283.5	1094.0	5 700	929.4	018 4	1013.0	1296.3	000 3	- 020	- 010 010	9 10.0 0 4 0 10	04210	0.0011			000.0	0.747	011011		0.700	000 1007		0.1021	1073.0	1048.0	1045.0
CBOD Influent mg/l	505.0 407.8	388.3	379.0	365.U	497.5	445.U	040.0 660.0	345.0	403.3	386.3	2000 207 S	395.6	543.8	477.0	0.75	345.0	0.005	0.025 0.006			2000			0.100	324.0	433.0		0.400		430.U	0.470 0010	207.0	303.0	422.0	3/8.0	330.0	352.0	350.0
Maximum Flow mgd	0.386 0.378	0.400	0.406	0.389	0.496	0.483	0.485	0.001	0.510			0.387	0.426			0.504	130.0		0.404	4/4/0 10/20	0.485	0.540	0.432	0.523	0.407	0.372	0.3/3	0.367	0.365	0.422	0.354	0.351	0.425	0.427	0.532	0.440	0.475	0.547
Average Flow mqd	0.331 0.316	0.286	0.321	0.326	0.364	0.375	0.408	0.426	0.390	0.404	0.360	0.347		0.205	0,210	0.218	0.363	0.342	0.373	0.403	0.401	0.404	0.357	0.322	0.283	0.313	0.304	0.306	0.313	0.320	0.300	0.264	0.326	0.334	0.397	0.390	0.357	0.358
Date	10/89 11/89	12/89	1/90	2/90	3/90	4/90	5/30	06/9		0/20 0/20	ne/e	10/30		08/21	1/21	19/2	12/5	4/91	19/3	6/91	7/91	8/91	9/91	10/91	11/91	12/91	1/92	2/92	3/92	4/92	5/92	6/92	7/92	8/92	9/92	10/92	11/92	12/92

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Dixon WWTP Upgrade Discharge Monitoring Report Summary Tabulation

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Chlorine Residual	0.10 0.08 0.16 0.09 0.09	3.75	0.1
С. С	$\begin{smallmatrix} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & & $	1141	25.9
Hd	7.5 7.7 7.4 7.6 7.5	339.6	7.7
SS Effluent mg/l	7.1 7.3 8.3 10.2 11.7	443.3	10.1
SS Effluent Ib/d	21.8 23.3 30.7 36.9 36.2	1282.4	29.1
CBOD Effluent mg/l	12.6 9.5 14.3 8.8	510	11.6
CBOD Effluent Ib/d	38.8 30.5 40.5 27.3	1467.5	33.4
SS Influent Ib/d	2320.0 603.5 983.0 772.8 738.0	41657.2	946.8
SS Influent mg/l	752.0 187.0 303.0 256.0 238.0	14291.3	324.8
CBOD Influent Ib/d	1391.0 805.2 901.0 772.8 874.0	47709.7	1084.3
CBOD Influent mg/l	451.0 249.5 278.0 256.0 282.0	16707.5	379.7
Maximum Flow mgđ	0.501 0.805 0.691 0.489 0.404	20.488	0.466
Average Flow mgd	0.370 0.387 0.389 0.362 0.362	15.264	0.347
Date	1/93 3/93 5/93 5/93	Total	Avg.

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### II. PLANT WASTEWATER TREATMENT PROCESS UNITS:

### A. Preliminary Treatment

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(1) Mechanical Bar Screen (Bar Screen Building)

In-channel type, stainless steel, front cleaning Bar rack width = 2 ft Total channel height = 5.0 ft Maximum flow depth at MDWF = 2 ft Bar thickness = 1/4" s.s Bar depth = 1 1/2"Clear spacing = 3/4"Clear opening efficiency = .75 Velocity through bar rack ADWF = .52 ft/s MDWF = 1.55 ft/s

(2) Bar Rack (Emergency Bypass Channel)

In-channel, manually cleaned Width of bar rack = 2 ft Depth at MDWF = 2 ft Bar thickness =  $\frac{1}{2}$ " Bar depth = 2" Clear spacing between bars = 1.0"

Clear opening efficiency

<u>1.0</u> = .67% .5+1.0

Velocity = Q/A = 4.64 cfs/(2x2x.67) = 1.73 ft/-S at MDWF= 1.55 cfs/(2x2x.67) = .58 ft/s at ADWF

#### B. Primary Settling Tank (Existing)

Configuration:

Diameter	45 ft
SWD	9.0 ft
Area	1590 sf
Volume	14,310 cf (107,031 gal)
Weir length	141.4 ft

Hydraulic Parameters:

Flow	Overflow Rat		stimated Removed North Strength Strengt	val We (SS) Loadin		Detention Time
MDWF	1887 gpd/sf	23%	, 4	0% 21,220	gpd/ft	0.94 hrs
ADWF	629 gpd/sf	36%	6	5% 7,070	gpd/ft	2.6 hr
Removals:						
<u>Flow</u>	Load	Raw	Ren	oval	<u>Effluent</u>	
MDW	F CBOD SS	3340 2920		16/d 8 16/d	2572 lb/d 1752 lb/d	
ADWI		3340 2920	lbs/d 120	2 1b/d 8 1b/d	2138 lb/d 1022 lb/d	
Primary Efflue	nt Recirculation:					
Pump	1 Characteristics:	Capacity TDH Motor	700 gpm 24.5 ft 7½ Hp, 115	) rpm		
Pump	2 Characteristics:	Capacity TDH Motor	1300 gpm 25 ft 15 Hp, 1150	) <del>r</del> pm		
Recirc	culation Ratio R =		/F 700 gpm/700 VF (700 + 1300	) gpm = 100% ) gpm/2083 = 96%		
С.	First Stage Tr	ickling Filter (I	Existing)			
	Configuration:	Diameter Stone Media Area Rock Volume	23			
	Hydraulic Para	meters & Remo	vals:			
	BOD Influent	Loading	Hydraulic Loading	Estimated Remov NRC	al Remova	al Effluer
MDWF	25 <b>72</b> lb/d 27	'1 lbs/d/1000 cf	1263 gpd/sf	60%	1543lb/	d 1029 lb/
ADWF	2138 lb/d 22	.5 lbs/d/1000 cf	421 gpd/sf	63%	1347lb/	d 791 lb

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NRC Removal Efficiency (%): = 100/(1+.0085(W/VF).<sup>5</sup>)

W = Influent BOD (lbs/d) V = Ac. ft Rock Volume F = 1 + R = 2R = Recirculation Ratio (100%)

### D. Second Stage Trickling Filter (Existing)

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Configuration:	Diameter	108 ft
U.	Stone Media Depth	6.5 ft
	Area	9160 sf (.21 ac)
	Rock Volume	59,540 cf (1.37 ac. ft)

Hydraulic Parameters & Removals:

	BOD Influent	Loading	Hydraulic Loading	Estimate Removal (NRC Formation)	Removal	Effluent
MDWF	10 <b>29</b> lb/d	17 lb/d/1000 cf	328 gpd/sf	68%	700 lb/d	329 lb/d
ADWF	791 lb/d	13 lb/d/1000 cf	109 gpd/sf	69%	546 lb/d	245 lb/d

NRC: Removal Efficiency (%) =  $100/(1+(.0085/1-E_1)(W_2/VF)^{.5})$ 

W = Influent BOD (lbs/d) V = Ac. ft Rock Volume F = 1 + R = 1.5R = Recirculation Ratio = Qr/Q = 50% of ADWF E<sub>1</sub> = Removal Efficiency through First Stage Filter

## E. Second Stage Setting Tank (Existing)

Configuration:	Diameter	34 ft
-	SWD	7 ft
	Area	908 sf
	Volume	6355 sf (47,532 gal)
	Weir Length	107 ft

Hydraulic Parameters:

Flow	Overflow Rate	Estimated (BOD	i Removal 9) (SS)	Weir Loading Rate	Detention Time
MDWF	3304 gpd/sf	(NRC)	33%	28,037 gpd/ft	.38 hr
ADWF	1101 gpd/sf	(NRC)	55%	9,346 gpd/ft	1.1 hr
	Removals:				
	Flow	Load	Raw	Removal	Effluent
	MDWF ADWF	CBOD SS CBOD SS	329 lb/d 1752 lb/d 245 lb/d 1022 lb/d	(NRC) (578) (NRC) 560 lb/d	329 lb/d 1174 lb/d 245 lb/d 462 lb/d

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### Third Stage Trickling Filter (Existing)

Configuration:	Diameter	55 ft
-	Stone Media Depth	4 ft
	Area	2376 sf (.054 ac)
	Rock Volume	9504 cf (.218 ac. ft)

Hydraulic Parameters & Removals:

	BOD Influent	Loading	Hydraulic Loading	Estimate Removal (NRC Formation)	Removal	Effluent	-
MDWF	329 lb/d	70 lb/d/1000 cf	1263 gpd/sf	49%	161 lb/d	168 lb/d	
ADWF	245 lb/d	56 lb/d/1000 cf	421 gpd/sf	52%	127 lb/d	118 lb/d	

NRC: Removal Efficiency (%) =  $100/(1 + (.0085/1-E_1) (W_2/VF)^{.5})$ 

W = Influent BOD (lbs/d)V = Ac.ft Rock VolumeF = 1 + R = 1.0R = No Recirculation

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### G. New Final Settling Tank

Diameter	65 ft
SWD	10.2 ft
Area	3318 sf
Volume	33,847 cf (253,154 gal)
Weir Length	204 ft
	SWD Area Volume

### Hydraulic Parameters:

Flow	Overflow Rate	Estimated (BOD)	Removal (SS)	Weir Loading Rate	Detention	
MDWF ADWF	904 gpd/sf 301 gpd/sf	(NRC) (NRC)	57% 75%	14,706 gpd/ft 4,902 gpd/ft	2 hrs 6 hrs	
<u>Flow</u>	Removals: Load	Raw	Removal	Plant Effluen	t NPDES Requirements	<u>.</u>
MDWF ADWF	CBOD SS CBOD SS	168 lb/d 1174 lb/d 118 lb/d 462 lb/d	(NRC) 669 lb/d (NRC) 347 lb/d	lbs/d       mg/l         168       6.7         505       20.2         118       14.1         115       13.8	* * * * 166.8 20	Ĺ

\*Maximum limits not known at this time

#### H. Chlorine Contact Tank (Existing)

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Configuration:	Tank Area Water Depth Tank Volume Detention Time MDWF ADWF	420 sf 8.5 ft 3570 cf (26,702 gal) *13.8 min *41.4 min
	Equipment Type Gas Storage	Vacuum Solution Feed By Gas 2 Ea 150 lb Cylinders in Heated Storage Building
	Storage Room Temperature Gas Delivery Capacity** Effluent Dosage Required Required Cl <sub>2</sub> Feed Capacity	65°F + 83 - 130 lb/d 10 mg/l at ADWF 83.4 lb/d

\*Includes volume in outlet structure and parshall flume only. Contact time in the effluent main to the Rock River outfall is not included in the detention time calculation

\*\*IEPA Standard = 41.5 lb/d per 150 lb cylinder at 65° F Manufacturer = 65 lb/d at l lb/day/°F

#### I. Parshall Flume

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Туре	Precast Fiberglass Flume & Inlet/Exit Channel Set in Cast in Place Monolithic Concrete
Flume Width	9"
Flow Measurement	0-5.7 MGD

## III. SLUDGE TREATMENT PROCESS UNITS (ANAEROBIC DIGESTION)

#### A. Sludge Production

Design Assumptions:

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- (1) All BOD removed converted to waste solids (No allowance for process destruction)
- (2) Effluent BOD = 20 mg/l; influent BOD = 400 mg/l
- (3) Volatile Suspended Solids (VSS) = 70% Total Solids (TS)
- (4) All waste sludge from primary treatment at 4% Avg. TSS Secondary & final stage sludges will be recirculated through the primary stage

BOD Removal Re	=	3170 lb/d		
		Removal Required	=	3170 lb/d
VSS Loading to I			=	2220 lb/d
Sludge Volume	-	3170/(.04x8.34)	-	9500 gpd
0	at 2% TS	3170/(.02x8.34)	=	19,000 gpd

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#### B. Primary Digester

Configuration:	Diameter	50 ft
U	SWD	19.4 ft
	Area	1963 sf
	Volume	38,023 cf (284,393 gal)
	Floating Steel	
	Duo Deck Cover Area	1925 sf

Total Deck Weight125,125 lbsFloating Cover Operating Pressure10.5" W.C.Gas System Operating Pressure9.5" W.C.Floating Cover Static Pressure12.5" W.C.

Digester Solids Loading = 2220/38.023 = 58 say 60 lb VSS/day/cf

#### Hydraulic Detention:

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Input Solids Concentration	Input Volume (gal/day)	Detention Time (Days)
2%	19,000	15
2.5%	15,204	18.7
3%	12,700	22.4
3.5%	10,860	26.2
4%	9500	30(Design)

## C. Digester Primary Sludge Feed Pump

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Pump Characteristics:

11" Plunger 5 Hp/230/460V/3ph Max. Capacity = 80 gpm TDH 75 ft

Sludge Loading	% Solids	Sludge Volume	Sludge Transfer Time
			80 gpm
3170 lb/d	4%	9500 gpd	2 hr
3170 lb/d	2%	19,000 gpd	4 hr

#### D. Digester Heating Requirement

Sludge Heating Requirement:

Qs = 19,000 gal sludge/day x 8.34 lb/gal x (95-55) °F x  $\frac{1 \text{ day }}{24 \text{ hr}}$  x  $\frac{1 \text{ Btu}}{16 \text{ °F}}$  = 264,100  $\frac{\text{Btu}}{\text{ hr}}$ 

Qs = 264,100 Btu/hr (at 2% TSS)

 $Q_s = 9500$  (8.34) (40) (1/24) (1) = 132,050 Btu/hr (at 4% TSS)

Assumptions:

- All walls buried to SWD
- Cover area = 1925 sf
- Bottom area = 1963 sf
- Wall area =  $\pi(50) \ge 21$  (full tank depth) = 3300 sf
- Digester operating temp = 95° F

**U-Factors** 

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Floating cover w/3" perlite insulation	U = .17
Walls on dry soil (due to sharply sloped raised berm)	U = .06
Bottom on wet soil (high water table condition)	U = .105

- Soil Temperature = 40° F
- Air Temperature =  $0^{\circ}$  F (cold climate winter conditions)

Heat flow equation through compound structures

 $Q = U \times A (T_2-T_1) \qquad Q = \text{Heat loss (Btu/hr)} \\ A = \text{Metered area normal to direction of flow (ft<sup>2</sup>)} \\ T_1 = \text{Outside temp (°F)} \\ T_2 = \text{Digester Op. temp (°F)} \end{cases}$ 

V = Heat transfer coeff. (Btu/sf.hr)

Heat Losses:

Cover  $Q_c = .17$  Btu x 1925 sf (95-0)°F= 31,089 Btu/hrsf.°F hrsf.°F hrWalls  $Q_w = .06 x 3300$  sf (95-40)= 10,890 Btu/hrFloor  $Q_f = .105 x 1963$  sf (95-40)= 11,336 Btu/hr $Q_t =$ Totai Heat Losses= 53,315 Btu/hr

Total Heat Requirement = Digester Sludge Heating Requirement  $(Q_s)$  + Total Heat Losses  $(Q_t)$ = 264,100 + 53,315 = \*317,415 Btu/hr

\*For worst conditions imposed on system (conservative)

#### E. Sludge Heater

Heat Requirement	317,415 Btu/hr
Heater Capacity	375,000 Btu/hr
Reserve Capacity	57,585 Btu/hr
Sludge Recirculation Rate	150 gpm
Boiler Output Rating	380,000 Btu/hr
Fired Surface	88 sf
Exhaust Fan Capacity	550 cfm
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## F. Fuel Requirements

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Furnace <u>Efficiency</u>	Fuel Energy Inj <u>Required (Btu/h</u>	
60%	529,025	At winter conditions & including cover & floor losses &
65%	488,331	Sludge at 2% TS
70%	453,450	
75%	423,220 (Desi	gn)
80%	396,769	

Check Heat Requirements For Anticiapted Actual Ambient Conditions:

	Summer	Winter
Raw Sludge From Primary	65°F	55°F
Air Temp	70°F	0°F
Soil Temp	55°F	40°F

Sludge at 4% TS; V sludge = 9500 gal/d

Summer Heat Requirements:

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			99,038 8,181 7,920 8,245
	Total Actual Summer Heat Requirement	=	123,384 Btu/hr
	Fuel Required at 75% Efficiency	-	164,512 Btu/hr
Winter	Heat Requirements:		
Walls	9500 (8.34) (95-55) x 1/24 x 1 .17 x 1925 (95-0) .06 x 3300 (95-40) .105 x 1963 (95-40)		132,050 31,089 10,890 11,336
	Total Actual Winter Heat Requirement	Ξ	185,365 Btu/hr
	Fuel Required at 75% Efficiency	=	247,153 Btu/hr

### Summary, Heat & Fuel Requirements:

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	Total Heat <u>Requirement (Btu/hr)</u>	Fuel Energy Requirement At 75% Efficiency (Btu/hr)
Worst Case Condition	317,415	423,220 (Design)
Anticipated Summer	123,384	164,512
Anticipated Winter	185,365	247,153

#### G. Gas Production Anaerobic Digestion

Assumptions:	40% Volatile Suspended Solids Destruction in digester (conservative) 15 cf of Gas Production/lb VSS destroyed 640 Btu/cf Heat Valve (methane gas) VSS P 70% of total solids
Gas Volume =	(3170 lb TS/d x .70 VSS/TS) x 40% Destruction x 15 $\underline{cf}$ Meth = 13,314 cf/day lb VS
Heat Value = 1	3,314 cf/d x 640 Btu/cf x $\frac{1d}{24}$ = 355,040 Btu/hr 24 hr

## H. Fuel Budget Balance (Btu/hr)

	Total Heat Requirement	Fuel Energy Requirement/75% Eff	Available Gas Energy	Excess (Flare Off)	Natural Gas Backup
Worst Case Condition	317,415	423,220	355,040		68,180
Anticipated Summer	123,384	164,512	355,040	190,528	
Anticipated Winter	185,365	247,153	355,040	107,887	

\* Under anticipated operating conditions, the digesters will generate required heating fuel. However, under worst case design conditions a natural gas backup will be required.

Natural Gas Backup Required Under Worst Case:

68,180 Btu/hr x 1 cf/1000 Btu Nat. Gas x 24 hr/day = 1640 cf Natural Gas/day

GNL-40-93

Digester Gas Flare Off Under Anticipated Conditions:

Summer 190,528 Btu/hr x 1cf/650 Btu x 24 hr/d = 7035 cf Dig. Gas/day Flare-off

Winter 125,340 (1/650) (24 hr/d) = 4628 cf Dig. Gas/day Flare-off

I. Digester Gas Flare

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Diameter 1234	1	
Capacity 2000	) cf/hr	
Flame Burn Duration:	Summer	3.5 hrs
	Winter	2 hrs

#### J. Digested Sludge Production

Raw Total Solids = 3170 lb/d (sludge input)

VS = 70% TS; 40% VSS Destruction

Total Sludge = 3170 lb/d 70% VSS 3170 (.7) = 2219 lb VSS/d 40% VSS Destruction Remaining VSS = (1-.4) 2219 = 1331.4 lb/day

Sludge Output = Total sludge - VSS Destruction = 3170 - (2219 - 1331.4)

> Sludge Output From Primary Digester to Secondary Digester = 2282 lb/d

Sludge Balance (Assume Digesters Concentrate Influent Sludge to 4% TS):

Input Sludge Concentration	Input Volume at 3170 lb/d (gal/day)	Output Volume at 2282 lb/d to Secondary (4% TS) (gal/day)	Supernatant Recycle to Head of Plant* (gal/day)
2%	19,005	6,841	12,164
2.5%	15,204	6,841	8,363
3%	12,700	6,841	5,859
3.5%	10,860	6,841	4,019
4.0%	9,503	6,841	2,662

\* Decant Draw Off By Gravity Flow to Head of Plant

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#### К. **Primary Sludge Recirculation (Pre-Heating)**

Pump Characteristics:	Capacity	150 gpm
	TDH	43 ft
	Motor	5 HP/1750 rpm

#### L. **Secondary Digester**

Configuration:	Diameter	37 ft
	SWD	18.5 ft
	Area	1075 sf
	Volume	19,891 cf (148776 gal)
	Floating Steel	1032 sf
	Duo Deck Cover Area	
	Operating Weight	49,100 lbs
	Gas Storage	7,575 cf
	Operating Pressure	9.5" W.C.
	Static Maximum Pressure	11" W.C.

Average Daily Gas Volume Production at ADWF & 3170 lb/d TS (3170 lb TS/d x 70 VSS/TS) x 40% Destruction x 15 cf Meth/lbVSS = 13,314 cf/d

Hourly Gas Production

 $= 555 \, \text{ft}^3/\text{hr}$ 

Storage Duration

13.7 hrs 57% of daily gas production

Sludge Transfer From Prin	nary 2282 lb/d <sup>7</sup>
Sludge Volume at 4% TS	6841 gal/d
Detention Time at	21.7 days

#### M. Secondary & Third Stage Sludge Recirculation

Secondary Recirculation:

Recirculation Ratio  $R = Q_R/Q = 50\%$  of ADWF = 350 gpm (.5 mgd)

Pump Characteristics:	Capacity	425 gpm
	TDH	32.5 ft
	Motor	5 HP/1750 rpm

Third Stage Recirculation:

Recirculation Ratio  $R = Q_R/Q = 50\%$  of ADWF = 350 gpm (.5 mgd)

Pump Characteristics:	Capacity	350 gpm
	TDH	24 ft
	Motor	5 HP/1750 rpm

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#### N. Sludge Transfer

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Primary to Secondary Digester or Secondary Digester to Dewatering Beds

Pump Characteristics:	Capacity	350 gpm
-	TDH	53 ft
	Motor	15 HP/1750 rpm

#### O. Sludge Dewatering Beds

(1) Existing Beds (to be rehabilitated):

Area5 ea at (26 x 100)= 13,000 sfVolume (1 ft depth)= 13,000 cf (97,233 gal)

- Type: Sand filter drain with concrete access pad Underdrain decant collection and recycle to 3rd Stage trickling filter influent
- (2) Proposed Additional Beds:

Area 6 ea at  $(29.2 \times 90)$  = 15,768 sf Volume (1 ft depth) = 15,768 cf (117,936 gal)

Type: Sand filter drain with concrete access driveway "strips". Underdrain decant collection and recycle to Bar screen influent channel.

#### (3) Annual Sludge Production:

2280 lb/d x 365 d/yr = 832,200 lb/yr (dry solids)

Volume at 4% solids = 2,494,604 gal/yr (liquid volume)

(4) Dewatering Capacity:

Total Bed Capacity = 97,233 + 117,936 = 215,170 gal

Desired Bed Turnover Level = 7 Turnovers/yr (52 days/turnover cycle)

Sludge Handling Required:

2,494,605 gal x 1/7 = 356,370 gal/bed turnover

% of required capacity available for de-watering =  $\frac{215,170}{356,370} = 60\%$ 

(5) Wet Haul Requirement (Tanker Truck):

40% of Annual Capacity = 997,840 gal/yr

Tanker Truck Discharge Outlet (gravity feed) to be provided at existing sludge beds

(6) Plant Sludge Storage Capacity:

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Primary Digester	284,393 gal
Secondary Digester	148,776 gai
Existing Sludge Beds	97,233 gal
New Sludge Beds	117,936 gal
Primary Sludge Pit	4,220 gal
*Total	652,558 gal

\* Pipe System Storage Not Included

Plant Design BOD Loading = 400 mg/l x 1 mdg x 8.34 = 3336 lb/d (conservative)

IEPA Design Criteria .17 lb/day/capita Per Capita Equivalent = 3336/.17 = 19,623 at proposed design plant loading

IEPA Design Criteria .09 lb/day/capita Sludge Production Plant BOD Loading = .09 x 19,623 = 1766 lb/day (by strict criteria)

Sludge Volume at 4% TS 1766 x (1/.04 x 8.34) x 365 d/yr = 1,932,324 gal/yr (5294 gpd)

Interim Sludge Storage Time:

652,558 gal/5294 gpd = 123 days

#### P. Dewatered Sludge Storage Pad

Daily Sludge Production 2282 lb/d

Assume Sludge Dewatered to 40% Solids Dewatered Sludge Density 45 lb/cf

Annual Dewatered Sludge Production = 2282 lb/d x 1/4 x 1 cf/45 lb x 365 d/yr = 46,274 cf (1714 cy)

Concrete Storage Pad Area5200 sfPush Wall Height5 ftRunoff Collection and Decant to 3rd Stage Trickling Filter Influent

## Q. Recommended Dewatered Sludge Land Application Equipment

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- (1) 4WD Tractor with Heavy Duty Front Loader, 22-25 cf Bucket Capacity
- (2) Push Cylinder Driven 220-340 cf Capacity Spreader, Tractor Pulled
- (3) Medium Capacity Dump Truck (tandem axle)

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60% of Annual Sludge Production Land Applied on Prison (State Owned) Cropland.

40% of Annual Sludge Production Contract Wet Hauled from Tanker Outlet

#### GENERAL

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- 0-2 IF DAMAGE IS DONE TO EXISTING OR NEW STRUCTURES DURING CONSTRUCTION, THEY SHALL BE REPLACED OR REPARED IN A SATISFACTORY MARKER BY THE CONTRACTOR AT HIS OWN EXPENSE.
- 6-1 STRUCTURAL SHEETS SHALL BE COORDINATED WITH ALL OTHER SHEETS FOR PPER SZZE AND LOCATIONS, ELECTRICAL REDUREMENTS, AND ANOHOR BO,TED ATLICAMENTS C-4.
- LOCATIONS AND ELEVATIONS OF EXISTING UNDEROROUND PIPING, C STRUCTURES, ETC., SHALL BE FIELD VERIFIED BY EXCAVATION (AS 0-5
- THE CONTRACTOR SHALL PROVIDE TEMPORARY SHORING AND BRACING FOR EXCAVITIONS AS REQUIRED BY ORMA AND ALL APPLICABLE BUILDING CODES. FAE CONTRACTOR IS RESPONSELE FOR THE ADEQUACY OF TEMPORARY SHORE AND BRACING FOR BUILDING AND EARTH LOADS.
- 0-5. NO PIPES OR SLIEVES FOR MECHANICAL TRADES SHALL PASS THROUG STRUCTURAL MEMBERS WITHOUT APPROVAL OF THE ENGINEER
- Performance Research Resource Performance 6-7
- EXISTING PIPING LOCATIONS AS DEPICTED ARE FOR GENERAL REFERENCE ONLY AND BASED ON RECORD INFORMATION FROM PREVIOUS CONSTRUCTION PROVIDE BY THE OWNER, IT IS ALSO PROMARE THE PLAN IS NOT Q-8.
- ALL PIPE LOCATIONS ARE TO BE FIELD VERIFIED PROR TO EXCAVATION. ALL ASSOCIATED REPAIR COSTS TO EXISTING KNOWN PIPPIG CAMAGED DERING EXEVATION WORK ARE THE RESPONSEDITY OF THE CONTRACTOR. RE-ROLTING OF DISTING LARGE THE ACCOMPOSITE NEW CONSTNUCTION ARE RE-ROLTING IN THE CONTRACT BIO PROCE AND WELL NOT RE SUBJECT TO C-8.

#### STRUCTURAL STEEL

- ALL STRUCTURAL STEEL SHALL BE ASTW A36 UNLESS NOTED OTHERWISE ALL WELDS SHALL BE MADE WITH ETDOX ELECTRODES AND SHALL CONFORM TO THE LATEST EDITION OF ANS SPECIFICATIONS. ALL WELDS SHALL BE POPOPAGE BY CORTINUED WELDRES.
- S-J. ALL STEEL BOLTS SHALL BE ASTM A325 UNLESS NOTED OTHERWISE
- LALESS SPECTIED OTHERWISE ON PLANS, STAINLESS STEEL BOLTS SHALL CONFORM TO THE FOLLOWING:
- SOLTS CONFORM TO ASTM F553 AND SHALL BE SUPPLIED WITH ONE WASHER AND NUT PER BOLT.
- 2. NUTS SHALL CONFORM TO ASTM \$54 3. WASHERS SHALL CONFORM TO ASTM & 240, TYPE 302

#### FOUNDATIONS

- THE CONTRACTOR SHALL FAMILIARIZE HIMSELF WITH THE SOL CONDITIONS IN THE FREAT AREA AND SHALL TAKE THE NECESISARY PRECANTIONS IN GROUP TO PROTECT DISTING THEORY FREM SHAREGE DUE TO HIS OPPERATIONS.
- The FOOTING LEVATIONS ARE BROWN ON THE FLANG. IF UNSUITABLE BEAMING REASE DEET AT THE RELATED FOOTING DEPTHS, THE FOUND-THIN BRULL REASEN DEPTHS TO THIN DEPTHS, THE FOUND-THIN BRULL REASEN DEPTHS TO THE REASEN DEPTHS TO THE REASEN DEPTHS THE REA F=2.
- F-3. ALL FOUNDATION EXCAVATIONS SHALL BE RESPECTED AND APPROVED BY THE ENGINEER PRIOR TO PLACEMENT OF CONCRETE.
- IT IS THE CONTRACTOR'S RESPONSIBILITY TO KEEP ALL STRUCTURA EXCAVATIONS DEWATCRED BURING CONSTRUCTION. THIS INCLUDES DEWATCRING WELLS AND YOR WELL POINTS IN RECEIVERARY.
- Involve of Four Inches of Crushed Stone (CA-6 or CA-16) Shall code involves the CA-16 contrast and edge S.Arbs attribut A+ found the Tre Crushed International Comparison (CA-6) and CA-16 STABLE SUBJECT NO COMPARED. The Comparison (S-16) and STABLE SUBJECT SUBJECT NO COMPARED. The Comparison (S-16) (S-16) and S-16) and S-16 and S-16 and S-16 and S-16 and S-16 and Counted Tools (S-16) and S-16 and S-16 and S-16 and S-16 and Counted Tools (S-16) and S-16 and S 5-5 HALL BE REPLACED BY 4" OF CONCRETE UN DESIMITY BY THE ENGINEER. A MINIMUM OF BITCHE COMPACTED TO BOX OF RELATIVE DE MARE SLAB AND FOOTINGS OF THE NEW CLAM
- HONT OF ANY FUL ALL WIGETARDAL INCLUDING THE ROOT I STRIPPED FROM THE STRIE. THE EXACT DETHS OF HE IS IN THE ALL WIGETARDAL HAS BEEN STRIPPED FROM SUBJECT STATE ALL WIGETARDAL HAS BEEN STRIPPED FROM SIGNICAS SHALL BE PROPERTIED WITH A FULLY COACED AREAS OF EXCESSIVE WIEDAKE, RUMPING, OR RUITING CHAIL REPLACED AND COMPLETE
- ALL FILL MATERIAL BHALL BE APPROVED FOR USE IN ADVANCE OF PLACEMENT BY THE DIGARDER. NO FILL SHALL BE PLACED OVER FROZEN, MUDOY OR THER DIGARDEROUS MATERIAL. NO FILL MAY BE FLACED OVER A PREVAUS UIT THAT HAS NOT BEEN ADEGUATELY COMPACTED AND ACCOPTED BY THE DIGARDER.
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#### CONCRETE

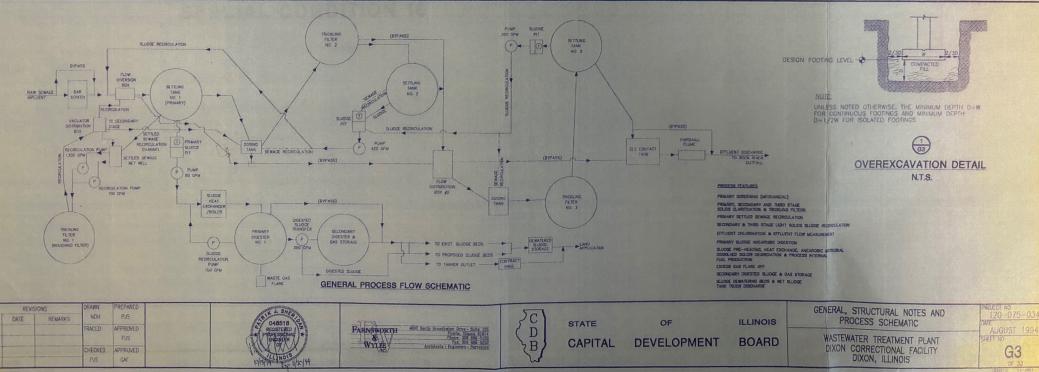
- ALL CONSTRUCTION ADMITS SHOWN ON THE DRAWINGS SHALL BE MODEPERATION THE STUTUES UNLESS TO ADMITLOS APPROVED BY THE CONNECTS REPRESENTATION OF ADMITS OF GONSTRUCTION ADMITS SHOLLES TO STAGENIATE CONSTRUCTION SHALL BE LOCATED AND DETARLS ON SHOP DRAWINGS. CONSTRUCTION ADMITS SHALL BE CONSTRUCTED AS FOLLOWS: C-1. THE REINFORCEMENT SHALL PASS CONTINUOUSLY THROUGH THE JOINT.
- 2 A KEY SHALL BE PROVIDED FOR ADEQUATE SHEAR TRANSFER AT ALL HORIZONTAL JOINTS. SHLAR KEYS SMALL BE AT LEAST 1 1/2 NOTES DEEP AND BE AT LEAST 1/3 OF THE WALL THROMESS AN WORL.
- VERTICAL JOINTS SHALL SE INTENTIONALLY ROUGHEDIED TO A 1/4 INCH MENDUM AMPLITUDE.
- WATESTOPS SHALL BE KADAGE TO CONSTRUCTION CONTS IN MATESTOPS THAT AND A CONSTRUCTION CONTS IN OF THE TYPE SHOWN ON CETAR 3 SHEET ST ALL WATESTOPS SHALL BE WIRED TO THE REDURCEMENT AS SHOWN ON DETAIL 35 WHEET ST. WATERSTOPS IN VIRTUAL JOINTS SHALL BE PLACED IN THE MODE OF THE WAIL ON THE SHALL BE PLACED IN THE MODE OF THE WAIL
- C-2 ALL CONCRETE SURFACES RECEIVEND A LAYER OF OROUT SHALL BE INTENTIONALLY ROUGHENED TO AN AMPLITUSE OF 1/2 IN. C-3.
- HERE LINE CONNETES IN O ME CAST AGAINET DESTRIO MACCIMUM ONE CONTENT IN CONTROL BURKLOC SHALL ME CAST, DET AND FREE OF LATANCE. A BOOMO ADDY SHALL ME LARD AT HE ADDY. THE BOOMO ADDY SHALL ME CAS DESTRIBUT SHOOTACHES. THE BOOMO ADDY SHALL ME LISED IN THE SECONDATIONES. THE BOOMO ADDY SHALL ME LISED IN STRUCT REDARDS TO POT LIFE COMPARENTLY WITH THE LISEDS IN WHO'N THE MAL HARE CONTENT. WITH THE LISEDS IN WHO'N
- C-4. THE CONCRETE STRENGTH, Fe. SHALL = 4000 PSI AT 28 DAYS C-8
- REINFORCEMENT BARS BIALL CONFORM TO THE REQUIREMENTS OF ASTM A-BIB. THE CONFRICT COVER OVER REINFORCEMENT SHALL BE 2° CEAR FOR CONFERT CAST ADMRST FORMS, NO 2° CLEAR FOR CONFERT CAST ADMRST CARTA, UNLESS OTHERWISE SHOW. DRACING OF REINFORCEMENTS BULLAURED FOM CONTENENT TO CENTERINE OF BARS. THE VIELD STRENGTH OF THE REINFORCEMENT FAULT BE REQUIRE FAULT

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- C-1. SIZE AND LOCATION OF CONCRETE BASES AND EMBEDDED INCHORACES FOR EQUIPMENT SHALL BE COORDINATED WITH THE COMPARENT SUPPLIER AND SHALL BE IN ACCORDANCE WITH APPROVED SUPPLIER AND SHALL BE IN ACCORDANCE WITH APPROVED SUCE DAMAGE.
- C-6 ALL DOWELS SHALL BE ANCHORED BITD PLACE BEFORE PLACEMENT OF CONCRETE. DO NOT "FLOAT" ANY "L" SHAPED DOWELS INTO PLACE

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- THE MASSHUM ALLOWABLE AGOREGATE SIZE FOR THE CONDRETE C-9.
- ALL REBAR SHOWN DOWELD INTO THE EXISTING CONCRETE ON THE FRAME BRALL BY INSTALLED IN ACCORDANCE WITH THE HIT C-TOS THSTIM BY HALL OR AN APPROVED EXIMALIST. THE LARREMONT LENGTH FOR THE REEMAN DOWELS INTO THE EXISTING CONCRETE SHALL BE AS FOLLOWS:
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- SAW CUTS THROUGH DOSTING CONCRETE SHALL BE " DEEP. DOSTING RENGREGARIUMT SHALL NOT BE CUT. EXISTING RENGREGARIUMT WHICH IS CALLED CUT FOR RELIES ON THE PLANS SHALL BE CLEANED TO RENGVE OLD CONCRETE. CIL
- C-12. CONCRETE COUPMENT PADS UP TO 12" IN TROCINEDS SHALL HAVE AND PLACED AT THE MID-HEIGHT OF THE CONCRETE PLACED
- C-13 ALL SAWGUT JOINTS SHALL BE FILLED IN WITH SKAFLEX-1A OR AN
- DODY ADRESSE AND HAR BOLTS SHALL BE TYPE 304 STANLISS THE AND SHALL BE NOTALIZE HARD DARKS BYPE THE HET HAR STOTE CHALL BE NOTALIZE HARD DARKS BYPE THE HARD STOTE CHAR BIR 222 (1988) TYPE A FRANC & 2 STARLSS STELL NUTE CHARDWORT LORGYS UNLESS OTHERWSE STARLSS STELL NUTE CHARDWORT LORGYS UNLESS OTHERWSE STARLSS STELL NUTE CHARDWORT LORGYS UNLESS OTHERWSE
- C-15. PUMP EQUIPMENT CONCRETE SUPPORT BASES TO EXTEND BEYOND EQUIPMENT SUPPORT BASE BY A MINIMUM OF 4 INCHES ON



## Special Condition 16.

- A. Subject to paragraph 8 below, an effluent limit of 0.5 mg/L Total Phosphorus 12 month rolling geometric mean (calculated monthly) basis (hereinafter "Limit"), shall be met by the Permittee by January 1, 2030, unless the Permitee demonstrates that meeting such Limit is not technologically or economically feasible in one of the following manners:
  - 1. The Limit is not technologically feasible through the use of biological phosphorus removal (BPR) process(es) at the treatment facility; or
  - 2. The Limit would result in substantial and widespread economic or social impact. Substantial and widespread economic impacts must be demonstrated using applicable USEPA guidance, including but not limited to any of the following documents:
    - a. Interim Economic Guidance for Water Quality Standards, March 1995 EPS-823-95-002;
    - b. Combined Sewer Overflows-Guidance for financial Capability Assessment and Schedule Development, February 1997 EPA-832-97-004;
    - c. Financial Capability Assessment Framework for Municipal Clean Water Act Requirements, November 2014; and
    - d. Any additional USEPA guidance on affordability issues that revises, supplements, or replaces those USEPA guidance documents; or
  - 3. The Limit can only be met by chemical addition for phosphorus removal at the treatment facility in addition to those processes currently contemplated; or
  - 4. The Limit is demonstrated not to be feasible by January 1, 2030, but is feasible within a longer timeline, then the Limit shall be met as soon feasible and approved by the Agency; or
  - 5. The Limit is demonstrated not to be achievable, then an effluent limit that is achievable by the Permittee (along with associated timeline) will apply instead except that the effluent limit shall not exceed 0.6 mg/L Total Phosphorus 12 month rolling geometric mean (calculated monthly).
- B. The Limit shall be met by the permittee by January 1, 2030, except in the following circumstances:
  - 1. If the Permittee develops a written plan, preliminary engineering report of the facility plan no later than January 1, 2030, to rebuild or replace the secondary treatment process(es) of the treatment facility, the Limit shall be met by December 31, 2035; or
  - 2. If the Permittee decides to construct/operate biological nutrient removal (BNR) process(es), incorporating nitrogen reduction, the Limit shall be met by December 31, 2035; or
  - If the Permittee decides to use chemical addition for phosphorus removal instead of BPR, and has a 1.0 mg/L Total Phosphorus monthly average shall be met by December 31, 2035; or
  - 4. If the Permittee has already installed chemical addition for phosphorus removal instead of BPR, and has a 1.0 mg/L Total Phosphorus monthly average effluent limit in its permit, or the Permittee is planning to install chemical addition with an IEPA construction permit that is issued on or before July 31, 2018, the 1.5 mg/L Total Phosphorus monthly average effluent limit (and associated compliance schedule) shall apply, and the Limit shall not be applicable.
  - 5. The NARP determines that a limit lower than the Limit is necessary and attainable The lower limit and timeline identified in the NARP shall apply to the Permittee.

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- 6. If the Permittee participates in a watershed group that is developing NARP for an impairment related to phosphorus or a risk eutrophication, and IEPA determines that the group has the financial and structural capability to develop the NARP by the deadline specified in the NARP provisions below.
- C. The Permittee shall identify and provide adequate justification of any exception identified in paragraph A or circumstance identified in paragraph B, regarding meeting the Limit. The justification shall be submitted to the Agency at the time of renewal of this permit or by December 31, 2035, whichever date is first. Any justification or demonstration performed by the Permittee pursuant to paragraph A or circumstance pursuant to paragraph B must be reviewed and approved by the Agency. The Agency will renew or modify the NPDES permit as necessary. No date deadline modification or effluent limitation modification for any of the exceptions or circumstances specified in paragraphs A or B will be effective until it is included in a modified or reissued NPDES Permit.
- D. For purposes of the permit the following definitions are used:
  - BPR (Biological Phosphorus Removal) is defined herein as treatment processes which do not require use of supplemental treatment processes at the treatment facilities before or after the biological system, such as but not limited to, chemical addition, carbon supplementation, fermentation, or filtration. The use of filtration or additional equipment to meet other effluent limits is not prohibited but those processes will not be considered part of the BPR process for purposes of this permit; and
  - 2. BNR (Biological Nutrient Removal) is defined herein as treatment processes used for nitrogen and phosphorus removal from wastewater before it is discharged. BNR treatment processes, s defined herein, do not require use of supplemental treatment processes at the treatment facility before or after the biological system, such as but not limited to chemical addition, carbon supplementation, formation or filtration. The us of filtration or additional equipment to meet other effluent limits is not prohibited, but those processes will not be considered part of the BNR process for purposes of this permit.
- E. The 0.5 mg/L Total Phosphorus 12 month rolling geometric mean (calculated monthly) effluent limit applies to the effluent from the treatment plant.

<u>SPECIAL CONDITION 17.</u> The Agency has determined that the Permittee's treatment plant effluent is located upstream of a waterbody or stream segment that has been determined to be at risk of eutrophication due to phosphorus levels in the waterbody. This determination was made upon reviewing available information concerning the characteristics of the relevant waterbody/segment and the relevant facility (such as quantity of discharge flow and nutrient load relative to the stream flow

A waterbody or segment is at risk of eutrophication if there is available information that plant, algal or cyanobacterial growth is causing or will cause violation of a water quality standard

The Permittee shall develop, or be a part of a watershed group that develops, a Nutrient Assessment Reduction Plan (NARP) that will meet the following requirements:

- A. The NARP shall be developed and submitted to the Agency by December 31, 2023. This requirement can be accomplished by the Permitee by participation in n existing watershed groups or by creating a new group. The NARP shall be supported by data and sound scientific rationale.
- B. The Permittee shall cooperate and work with other stakeholders in the watershed to determine the most cost-effective means to address the risk of eutrophication. If other stakeholders in the watershed will not cooperate in developing the NARP, the Permittee shall develop its own NARP for submittal to the Agency to comply with this condition.
- C. In determining the target levels of various parameters necessary to address the risk of eutrophication, the NARP shall either utilize the recommendations by the Nutrient Science Advisory Committee or develop its own watershed specific target levels.
- D. The NARP shall identify phosphorus input reductions from point sources and non-point sources in addition to other measures necessary to remove the risk of eutrophication characteristics that will cause or may cause violation of a water quality standard. The NARP may determine, based on an assessment of relevant data, that the watershed does not have a risk of eutrophication related to phosphorus, in which case phosphorus input reductions or other measures would not be necessary. Alternatively the NARP could determine that phosphorus input reduction from point sources are not necessary, or that phosphorus input reductions from both point and nonpoint source are necessary, or that phosphorus input reductions are not necessary and that other measures, besides phosphorus input reductions are necessary.
- E. The NARP shall include a schedule for the implementation of the phosphorus input reductions and other measures. The NARP schedule shall be implemented as soon as possible and shall identify specific timelines applicable to the permittee.
- F. The NARP can include provisions for water quality trading to address the phosphorus related risk of eutrophication characteristics in the watershed. Phosphorus/Nutrient trading cannot result in violations of water quality standards or applicable antidegradation requirements.
- G. The Permittee shall request modification of the permit within 90 days after the NARP has been completed to include necessary phosphorus input reductions identified within the NARP. The Agency will modify the permit if necessary.
- H. If the Permittee does not develop or assist in developing the NARP and such a NARP is developed for the watershed, the Permittee will become subject to effluent limitations necessary to address the risk of eutrophication. The Agency shall calculate these effluent limits by using the NARP and any applicable data. If no NARP has been developed the effluent limits shall be determined for the Permittee on a case-by-case basis, so as to ensure that the

Permittees discharge will not cause or contribute to violations of the dissolved oxygen or narrative offensive condition of water quality standards;



April	2.10
May	2.10
June	1.80
July	2.20
August	2.00
September	2.00
October	2.40
November	2.40
December	2.40
January	2.50
February	1.70
March	2.10
GEOMEAN	2.13



2323 Fourth Street P.O. Box 483 Peru, Illinois 61354 815-224-1650 800-659-4659 FAX 815-224-1688 www.testinc.com

## **ANALYTICAL REPORT**

June 20, 2023

David Tullerow Attn: David Tullerow 215 W. Chilppewa St Dwight, IL 60420

Order Number: 230600301

TEST, Inc. received 1 sample on June 08, 2023 at 10:45 for the analyses presented in the following report.

There were no problems with the analyses and all the data for the associated QC met EPA or laboratory specifications . Exceptions would be described in the Case Narrative, if applicable.

This final Analytical Report consists of this cover letter, case narrative, laboratory results and any accompanying documentation including, but not limited to, chain of custody records.

This report may not be reproduced, except in full, without prior written approval from TEST, Inc.

If you have any questions regarding these test results, please do not hesitate to contact me at (815) 224-1650 or (800) 659-4659.

Sincerely,

Total Environmental Service Technologies, Inc.

Layne Zene

Layne Zens Quality Assurance Officer



Page 1 of 4

CLIENT: David Tullerow

DATE: June 20, 2023

ORDER NUMBER: 230600301



2323 Fourth Street P.O. Box 483 Peru, Illinois 61354 815-224-1650 800-659-4659 FAX 815-224-1688 www.testinc.com

GENERAL COMMENTS:

All results reported in wet weight unless otherwise indicated (mg/kg = Dry Weight).

Sample results relate only to the analytes of interest and to the sample as received by the laboratory.

Accreditation by the State of Illinois is not an endorsement or a guarantee of the validity of data generated. For more information about the laboratories' scope of accredition in regards to Alkalinity/SM2320B; Chloride/SM4500Cl-B; TDS/SM2540C; Fluoride/SM4500F-C; Nitrate/EPA353.2R2; Nitrite/SM4500NO2-B; pH/SM4500H-B; Sulfate/ASTM D516-90; Al, Ba, Cr, Cu, Fe, Mg, Mn, Ni, Ag, Na, Zn/EPA 200.7R4.4; As, Be, Cd, Cr, Cu, Pb, Mn, Ni, Se, Ag/EPA200.8R5.4 please contact TEST, Inc. or the Agency.

LIMIT: Method Detection Limit. The minimum concentration of an analyte that can be measured and reported with a 99% confidence that analyte is greater than zero.

Outside laboratories references under analyst on the Final Report include:

PACE: Pace Analytical QCA: QC Analytical, LLC SB: Suburban Laboratories ESI: EnviroScience, Inc. EURO: Eurofins TEK: Teklab, Inc.

The subcontracted analytical report (if not provided) is available upon request.

#### DATA QUALIFIERS:

- A: Sample was received in lab with improper preservation
- B: Analyte detected in the associated Method Blank

J: The analyte was positively identified; the associated numerical value is less than the RL but greater than or equal to the MDL, and the concentration is an approximate value

TNTC: Too Numerous to Count

- H: Analysis run past method hold time
- I: Invalid
- L: Result was over the maximum contaminant level set by the EPA
- ND: Non-Detect
- T: Sample received outside thermal preservation acceptance criteria
- S: Sample sent to subcontracted NELAP Laboratory
- Q: Results accepted outside of quality control limits

#### METHOD REFERENCES:

EPA: USEPA Methods for the Determination of Inorganic Substances in Environmental Samples; Methods for Chemical Analysis of Water and Wastes; Methods for Organic Chemical Analyses of Municipal and Industrial Wastewater; 40CFR136 App A; Methods for Determination of Metals in Environmental Samples; Methods for the Determination of Organic Compounds in Drinking Water

SW: USEPA, Test Methods for Evaluating Solid Waste, SW-846, 3rd Ed, includes Updates I-III

SM: Standard Methods for the Examination of Water and Wastewater, 20th Ed. D: ASTM, Annual Book of Standards

Reported To:

David Tullerow David Tullerow 215 W. Chilppewa St Dwight, IL 60420 Order No.: 230600301 P.O. No.: Date Received: 06/08/2023 Collected by: David Tullerow Report Date: 06/20/2023 PWS ID No.:

## Laboratory Results

		Location:	WW INF
Date Collected: 06	/08/2023	Туре:	Grab
Time Collected: 09	:45	Sample Matrix:	Waste Water

Laboratory Test	Tested Value	Test Units	Detection Limit	Qualifier	Test Method	Date of Analysis	Analyst
Phosphate,Total	6.57	mg/L	0.614		SM4500PBE	6/19/23 13:	00 JN

Cash	Specify Regulatory Program:	None (info only)	503 Sludge   *Other	* Please specify in sampler remarks	section below.									_	16COW	EWA STWI	60220	Atton Com		2000			
<b>Y RECORD</b> RU, IL 61354• (815) 224-1650	Analysis Required			sade										Sampler Remarks:	DAVID TUTIEROW	215 W. CHIPPENASTWI	DWIGHT IL	auto course cierus	Lab Remarks:				
CHAIN OF CUSTODY RECORD TEST, INC. • 2323 4TH STREET • P.O. BOX 483 • PERU, IL 61354 • (815) 224-1650	Client Name:	-	email address:	Preservative pH Matrix Qty Sampling Location	ww inf									2 7 Received By: Date: (0/8/23	Received By: Date: / _ /	Received By: Date: / /		Received By: Date: _/ _/	Aluo	Yes No□ Temperature of samples: / 2 °C	No <sup>1</sup> Sample containers intact ?: Yes	Not Sumicient sample volume 2: Yesty Not Not COC properly completed 7: Yesty Not	ſ
T (In C.)	Project #	hature:		Samp Collection Time Grab Pre	1 6/8/23 9 the X	3 / /	4 / / :	5 / / :	9 : : :	7 / / :	8 / / 8	: / / 6	10 / / :	Relinquistred By: Lo S / S / Time: Lo S / S / Time: Lo S / S /	Relinquished By:	Relinquished By: Date: / /		Reinquished By: Date: _/_/			Samples in proper container / bottle ?: Yes		